

ANALYTICAL REPORT

METHANE AND CARBON DIOXIDE ADSORPTION ISOTHERMS

LAST CHANCE #3

**PREPARED FOR
GASSY COAL COMPANY NL**

APRIL 2050



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ADSORPTION ISOTHERM ANALYSIS

INTRODUCTION

Twelve samples were received from the Gassy Coal Company NL Last Chance #3 well (see table below) to be evaluated for adsorption isotherm properties (composites only), vitrinite reflectance (composites only) and maceral content (all samples). Adsorption isotherm data are reported herein.

Last Chance #3 Work Programme

Sample No.	From (m)	To (m)	Mass (g)	Isotherm No.	Temp. (°C)	Gases					Petrographics	
						CH ₄	C ₂ H ₄	C ₃ H ₆	N ₂	CO ₂	Maceral	Reflectance
A	526.04	526.78	2616	n/a	n/a	n/a	n/a	n/a	n/a	n/a	Y	N
B	557.43	557.81	4040	Iso. 1	54.5	Y	N	N	N	Y	Y	Y
C	558.91	559.77	3324									
D	563.64	564.02	2456									
E	565.96	566.91	3442									
F	569.41	570.36	3134									
G	619.48	619.83	1930	n/a	n/a	n/a	n/a	n/a	n/a	n/a	Y	N
H	620.12	621.03	1342	n/a	n/a	n/a	n/a	n/a	n/a	n/a	Y	N
I	621.36	622.29	3702	n/a	n/a	n/a	n/a	n/a	n/a	n/a	Y	N
J	652.08	653.05	3602	n/a	n/a	n/a	n/a	n/a	n/a	n/a	Y	N
K	663.83	664.33	2518	n/a	n/a	n/a	n/a	n/a	n/a	n/a	Y	N
L	667.93	668.43	2054	n/a	n/a	n/a	n/a	n/a	n/a	n/a	Y	N

ADSORPTION ISOTHERMS

Sample Preparation

Individual samples were received as -6 mm coal which had been crushed elsewhere. Approximately 200 g of each sample was received. Composites were made by weighting each sample by mass of coal before combining the individual samples to produce a composite of total weight of 150 g. This 150 g sample was sub-sampled to obtain around 25 g for petrographic analysis. The remaining sample was then crushed to less than 0.212 mm and brought to an equilibrium moist state (AS 1038.17-1989) for methane adsorption isotherm analysis.

Moisture Content and Ash Yield

Equilibrium moisture content, ash yield and helium density (for evaluation of dead volume of the system) were determined prior to methane adsorption isotherm analysis.

Moisture content was determined by weighing approximately 0.5 to 1 g of coal in a 4 digit balance (i.e to 0.1 mg), heating it to 110 °C in a nitrogen atmosphere, and then re-weighing the dry sample. The method is similar to AS1038.3-1989 "Methods for the analysis and testing of coal and coke. Part 3: Proximate analysis of higher rank coal."

Ash yield was determined on dry coal by weighing approximately 0.5 to 1 g of dry coal in a 4 digit balance (i.e. to 0.1 mg), heating it to 500 °C and holding it there for one hour and then raising the heat to 815 °C and maintaining this for a further one hour. The ash yield is then determined following re-weighing at room temperature. The method is similar to AS1038.3-1989 "Methods for the analysis and testing of coal and coke. Part 3: Proximate analysis of higher rank coal."

Adsorption Isotherm

Adsorption isotherm determination used at least nine pressure steps up to a maximum pressure of around 8 MPa (1160 psia) where the coals are at 800 m or shallower and additional pressure steps may be used for deeper coals. Analysis was performed as close as possible to formation temperature as advised. The procedure is detailed as follows:

1. adsorption isotherm cells are weighed
2. the equilibrium moist coal is placed in the cells and the cells weighed
3. the cells are evacuated and weighed
4. helium is introduced into the cells at pressures of approximately 2, 4, 6 and 8 MPa; this data is used to calculate the free volume of the cells and consequently the helium density of the coal
5. the cells are evacuated
6. a fixed volume of methane is introduced into the cell and the pressure monitored to the nearest 1 kPa until there is no change in pressure for a period of at least one hour
7. the adsorption is determined
8. steps 6 and 7 are repeated for each pressure step

Pressure steps for test gases are nominally at 250, 500, 1000, 2000, 3000, 4000, 6000 and 8000 kPa but may vary according to gas and sample depth. A test pressure of 8000 kPa is equivalent to approximately 800m hydrostatic depth.

Note Samples are retained by Energy Resources Consulting Pty Ltd for a period of 12 months following the reporting date. They may be discarded after this time.

RESULTS

Adsorption isotherm results are tabulated and presented graphically as follows. Results are also provided as Excel spread sheet data.

Absolute adsorption isotherms are calculated. Results are presented at standardised conditions of 20°C and 1 atmosphere (101.3 kPa) pressure per gram of coal (which is equivalent to m³/t) as well as in standard cubic feet (scf) at 60 °F and 1 atmosphere (14.7psia) per short ton (2000 lb) of coal.

Gibbs (excess) adsorption and other reporting bases (pressure / temperature) or as mmol/g can also be reported on reported on request.

Single Gas Adsorption Isotherm Modelling

Single gas adsorption isotherms are modelled according to Langmuir for both as analysed and daf (dry, ash free) bases:

$$V = \frac{V_L P}{(P + P_L)}$$

where V = volume adsorbed

P = pressure

V_L = Langmuir volume constant

P_L = Langmuir pressure constant

Data is presented graphically as both the Langmuir plot and as a plot of P versus P/V which shows the goodness of fit of the data to the model.

Extended Langmuir Modelling for Gas Mixtures

Both methane and carbon dioxide adsorption isotherms were determined. The adsorption isotherm of gas mixtures can be assessed using the extended Langmuir model:

$$V_i = \frac{V_{Li} P y_i}{P_{Li} \left(1 + \sum_{j=1}^n \frac{1}{P_{Lj}} y_j P \right)}$$

where :

V_i = storage capacity of component i (cc/g or m³/t or scft/t)

P_{Li} = Langmuir pressure of the ith phase obtained from the single component data

V_{Li} = Langmuir volume of the ith phase obtained from the single component data

P = pressure (kPa or psi)

y_i = the mole fraction of species i in the free gas phase

n = number of components

This modelling has been carried out for a mixture of CH₄:CO₂ = 50:50 for the free gas phase.

Other Adsorption Isotherm Models

BET, Dubinin-Radushkevich (DR) and Dubinin-Astakhov (DA) modelling may also be undertaken:

BET:
$$\frac{1}{V \left\{ \left(\frac{p^0}{p} \right) - 1 \right\}} = \frac{1}{V_m C} + \frac{(C-1)P}{V_m C p^0}$$

Dubinin-Radushkevich (DR):
$$\log V = \log V_0 - D \left[\log \left(\frac{p^0}{p} \right) \right]^2$$

Dubinin-Astakhov (DA):
$$\log V = \log V_0 - D \left[\log \left(\frac{p^0}{p} \right) \right]^n$$

Constants calculated for these models are tabulated.

Methane (CH₄) Adsorption

Client	Gassy Coal Company NL
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Sample Details	Last Chance #3 - Iso 1 (557.43 - 570.36 m)
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Sample Properties			
Inherent Moisture (% ad)	0.7	Isotherm Sample Mass (g) [lb]	122.47 [0.27000]
Ash (% ad)	42.4	Particle Size (mm) [US mesh]	-0.212 [70]
Volatile Matter (% ad)	14.4	Helium density (g/cc)	1.681
Fixed Carbon (% ad)	42.5	Test Temperature (°C) [°F]	54.5 [130.1]
Ash (% Equilibrium Moisture ba)	42.1	Analysis date	25-Feb-50
Moist. (% Equilibrium Moisture)	1.6	Test Gas	Methane

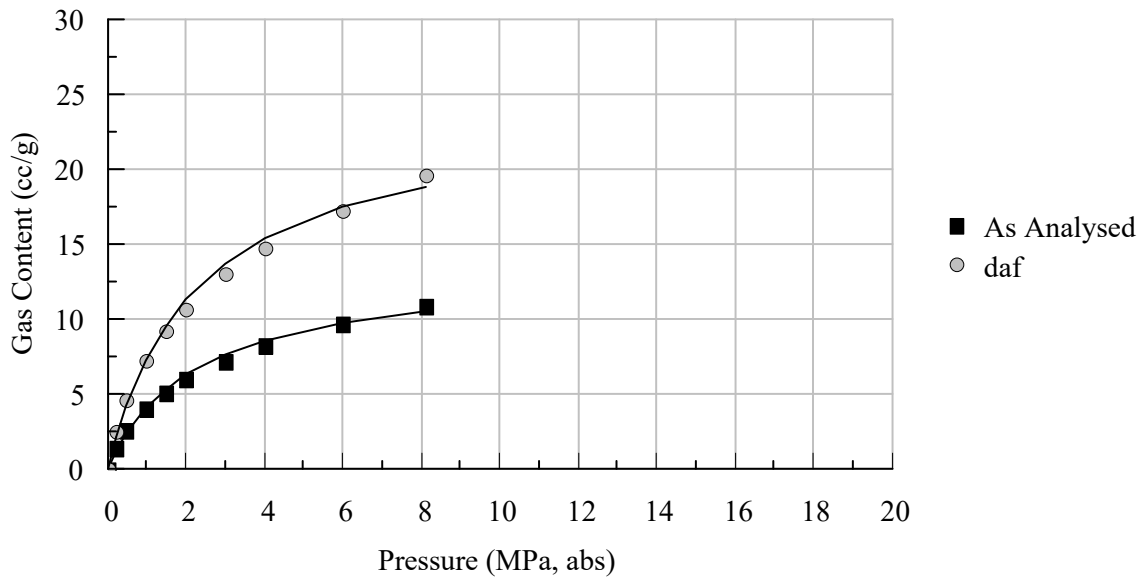
Methane Absolute Adsorption at Equilibrium Moisture Basis

at 20°C; 101.1kPa (1 atm)			at 60°F, 14.7 psia		
Pressure (MPa) (absolute)	Gas Content (cc/g) (as analysed)	Gas Content (cc/g) (daf)	Pressure (psia)	Gas Content (scf/t) (as analysed)	Gas Content (scf/t) (daf)
0.234	1.47	2.61	34	46	83
0.501	2.64	4.69	73	83	148
1.005	4.14	7.36	146	131	232
1.494	5.20	9.23	217	164	291
2.018	6.07	10.78	293	192	340
3.015	7.35	13.05	437	232	412
4.015	8.35	14.83	582	263	468
5.993	9.78	17.37	869	309	548
8.130	11.04	19.61	1179	348	619

Langmuir Isotherm Coefficients

	P_L (MPa, abs)	V_L (cc/g)	P_L (psia)	V_L (scf/t)
As analysed	2.31	13.67	335	431
daf	2.31	24.28	335	766

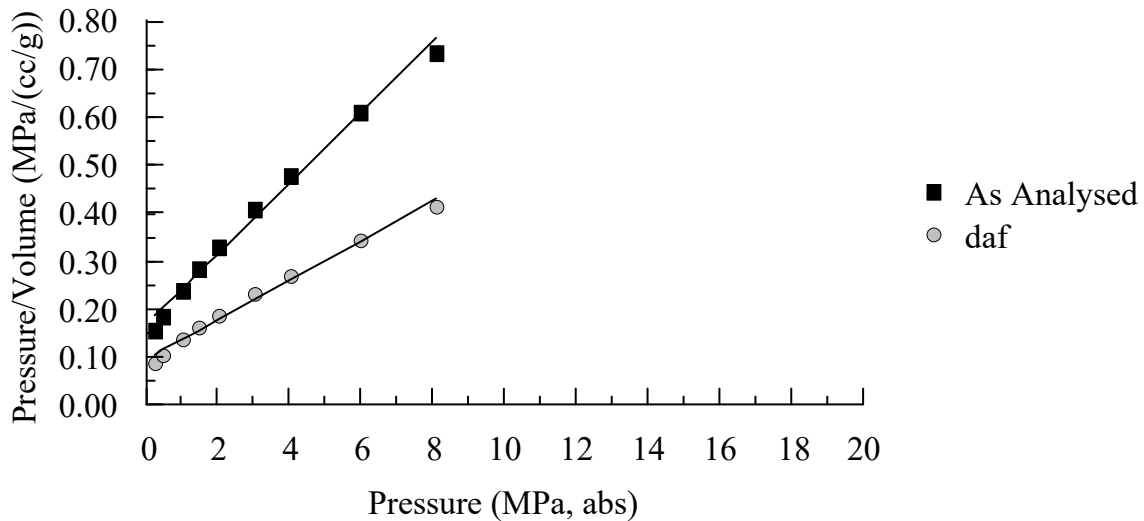
Methane Absolute Adsorption Isotherm (20 °C, 101.3 kPa)
 Last Chance #3 - Iso 1 (557.43 - 570.36 m)
 Analysis Temperature 54.5 °C



$$V = 13.67 P / (P + 2.31) \quad (\text{As analysed})$$

$$V = 24.28 P / (P + 2.31) \quad (\text{daf})$$

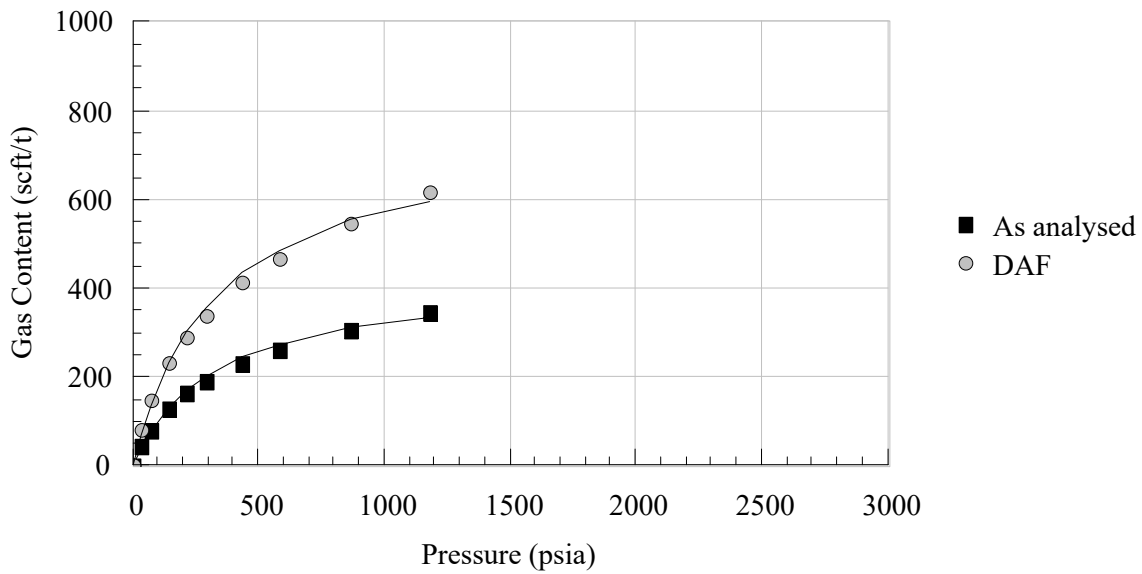
Methane Absolute Adsorption Isotherm (20 °C, 101.3 kPa)
 Last Chance #3 - Iso 1 (557.43 - 570.36 m)
 Analysis Temperature 54.5 °C



$$P/V = 0.073 P + 0.169; \quad r^2 = 0.991 \quad (\text{As analysed})$$

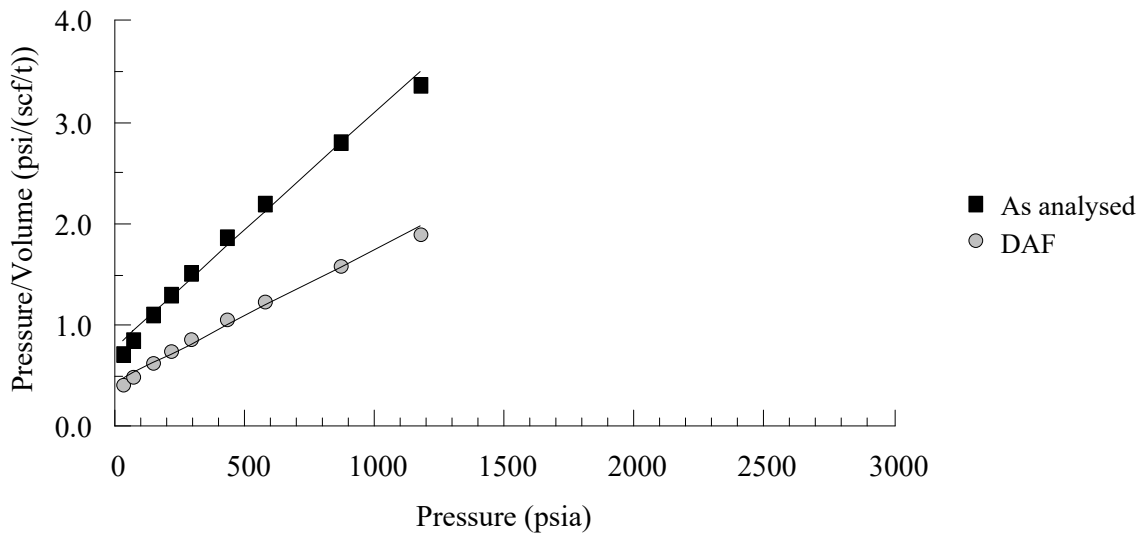
$$P/V = 0.041 P + 0.095; \quad r^2 = 0.991 \quad (\text{daf})$$

Methane Absolute Adsorption Isotherm (60 °F, 14.7 psia)
 Last Chance #3 - Iso 1 (557.43 - 570.36 m)
 Analysis Temperature 130.1 °F



$V = 431 P / (P + 335)$ (As analysed) $V = 766 P / (P + 335)$ (daf)

Methane Absolute Adsorption Isotherm (60 °F, 14.7 psia)
 Last Chance #3 - Iso 1 (557.43 - 570.36 m)
 Analysis Temperature 130.1 °F



$P/V = 0.00232 P + 0.776$; $r^2 = 0.991$ (As analysed)
 $P/V = 0.00130 P + 0.437$; $r^2 = 0.991$ (daf)

Client	Gassy Coal Company NL
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Sample Details	Last Chance #3 - Iso 1 (557.43 - 570.36 m)
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Sample Properties			
Inherent Moisture (% ad)	0.7	Isotherm Sample Mass (g) [lb]	122.47 [0.27000]
Ash (% ad)	42.4	Particle Size (mm) [US mesh]	-0.212 [70]
Volatile Matter (% ad)	14.4	Helium density (g/cc)	1.681
Fixed Carbon (% ad)	42.5	Test Temperature (°C) [°F]	54.5 [130.1]
Ash (% Equilibrium Moisture ba)	42.1	Analysis date	25-Feb-50
Moist. (% Equilibrium Moisture)	1.6	Test Gas	Methane

Methane Absolute Adsorption at Equilibrium Moisture Basis

Model Co-efficients					
		P in MPa, abs V in cc/g		P in psia V in scf/t	
		aa	daf	aa	daf
Langmuir Coefficients					
	PL	2.31	2.31	335	335
	VL	13.67	24.28	766	766
model fit :		no. of points = 9		r ² = 0.9907	
BET Coefficients					
	V _m	10.23	18.17	n/a	n/a
	C	27.44	27.44	n/a	n/a
model fit :		no. of points = 9		r ² = 0.9992	
D-R Coefficients					
	V _o	13.25	23.53	n/a	n/a
	D	0.19	0.19	n/a	n/a
model fit :		no. of points = 9		r ² = 0.9990	
D-A Coefficients					
	n	1.81	1.81	n/a	n/a
	V _o	14.50	25.76	n/a	n/a
	D	0.23	0.23	n/a	n/a
model fit :		no. of points = 9		r ² = 0.9998	
po = 42.356					

Carbon Dioxide (CO₂) Adsorption

Client	Gassy Coal Company NL
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Sample Details	Last Chance #3 - Iso 1 (557.43 - 570.36 m)
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Sample Properties			
Inherent Moisture (% ad)	0.7	Isotherm Sample Mass (g) [lb]	130.54 [0.28779]
Ash (% ad)	42.4	Particle Size (mm) [US mesh]	-0.212 [70]
Volatile Matter (% ad)	14.4	Helium density (g/cc)	1.720
Fixed Carbon (% ad)	42.5	Test Temperature (°C) [°F]	54.5 [130.1]
Ash (% Equilibrium Moisture ba)	42.1	Analysis date	25-Feb-50
Moist. (% Equilibrium Moisture)	1.4	Test Gas	Carbon Dioxide

Carbon Dioxide Absolute Adsorption at Equilibrium Moisture Basis

at 20°C; 101.1kPa (1 atm)

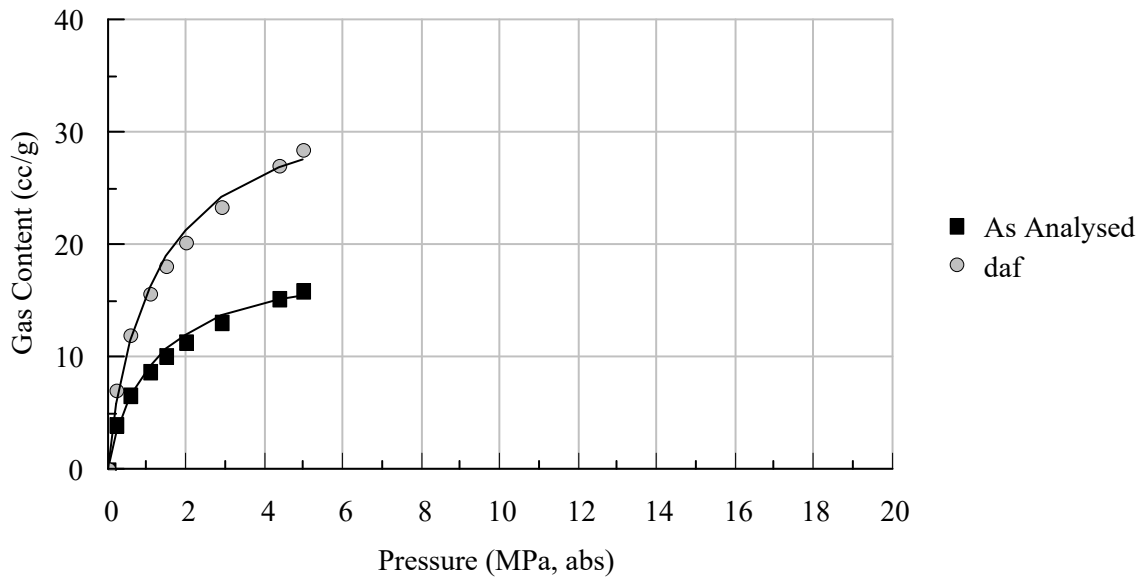
at 60°F, 14.7 psia

Pressure (MPa) (absolute)	Gas Content (cc/g) (as analysed)	Gas Content (cc/g) (daf)	Pressure (psia)	Gas Content (scf/t) (as analysed)	Gas Content (scf/t) (daf)
0.244	4.04	7.15	35	128	226
0.605	6.79	12.01	88	214	379
1.078	8.92	15.79	156	282	498
1.494	10.23	18.10	217	323	571
1.976	11.41	20.19	287	360	637
2.924	13.19	23.35	424	416	737
4.390	15.36	27.18	637	485	858
4.998	16.06	28.43	725	507	897

Langmuir Isotherm Coefficients

	P_L (MPa, abs)	V_L (cc/g)	P_L (psia)	V_L (scf/t)
As analysed	1.18	19.30	172	609
daf	1.18	34.15	172	1078

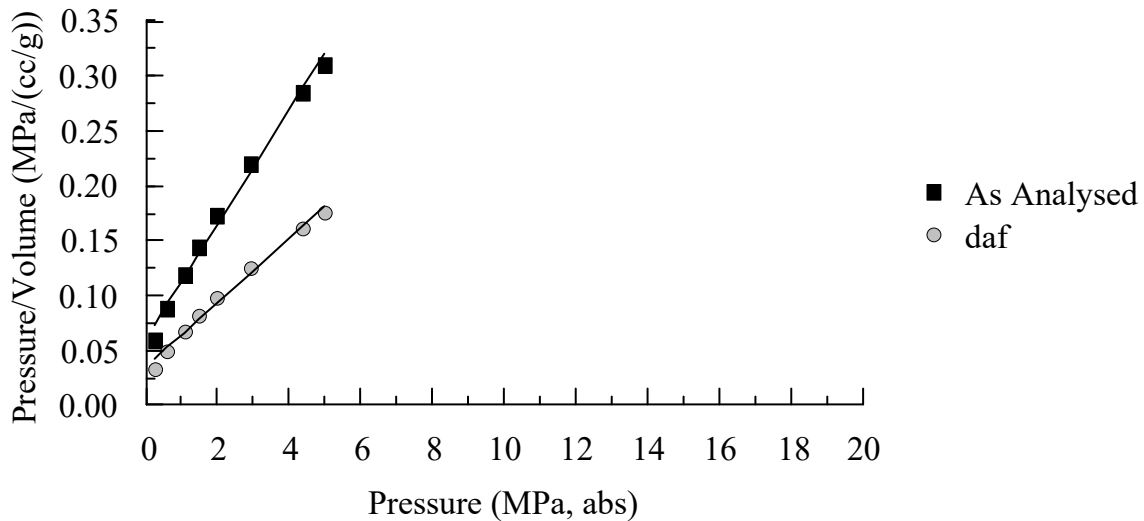
Carbon Dioxide Absolute Adsorption Isotherm (20 °C, 101.3 kPa)
 Last Chance #3 - Iso 1 (557.43 - 570.36 m)
 Analysis Temperature 54.5 °C



$$V = 19.30 P / (P + 1.18) \quad (\text{As analysed})$$

$$V = 34.15 P / (P + 1.18) \quad (\text{daf})$$

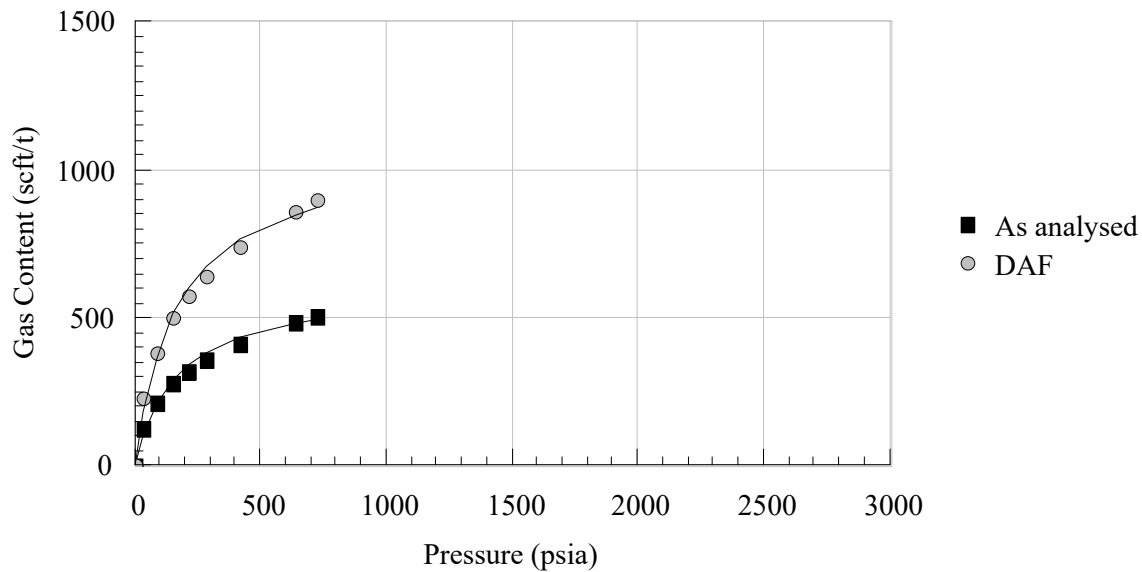
Carbon Dioxide Absolute Adsorption Isotherm (20 °C, 101.3 kPa)
 Last Chance #3 - Iso 1 (557.43 - 570.36 m)
 Analysis Temperature 54.5 °C



$$P/V = 0.052 P + 0.061; \quad r^2 = 0.991 \quad (\text{As analysed})$$

$$P/V = 0.029 P + 0.035; \quad r^2 = 0.991 \quad (\text{daf})$$

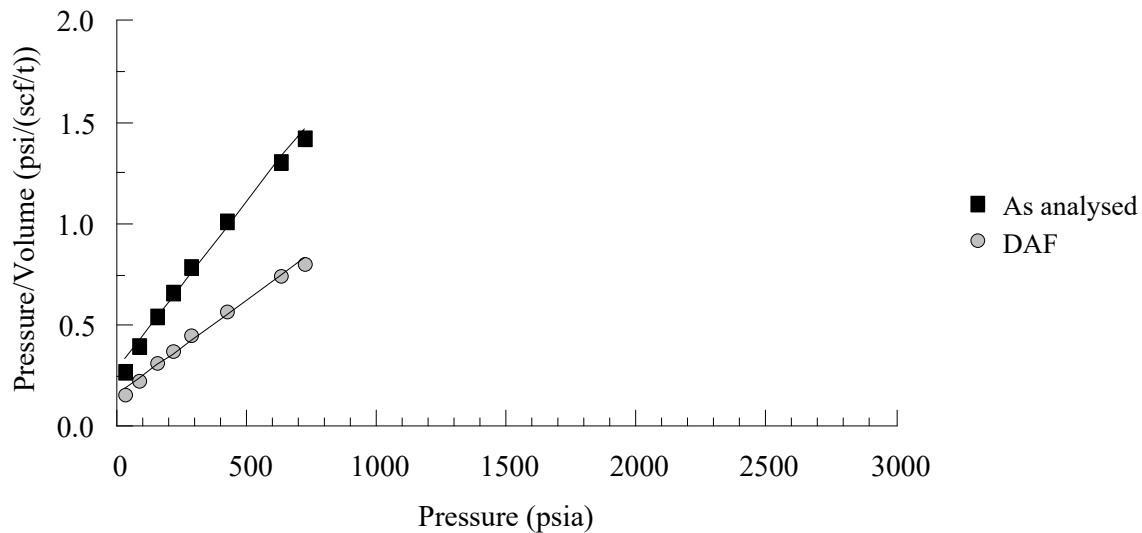
Carbon Dioxide Absolute Adsorption Isotherm (60 °F, 14.7 psia)
 Last Chance #3 - Iso 1 (557.43 - 570.36 m)
 Analysis Temperature 130.1 °F



$$V = 609 P / (P + 172) \quad (\text{As analysed})$$

$$V = 1078 P / (P + 172) \quad (\text{daf})$$

Carbon Dioxide Absolute Adsorption Isotherm (60 °F, 14.7 psia)
 Last Chance #3 - Iso 1 (557.43 - 570.36 m)
 Analysis Temperature 130.1 °F



$$P/V = 0.00164 P + 0.282; \quad r^2 = 0.991 \quad (\text{As analysed})$$

$$P/V = 0.00093 P + 0.159; \quad r^2 = 0.991 \quad (\text{daf})$$

Extended Langmuir Modelling

Client	Gassy Coal Company NL
Sample Details	Last Chance #3 - Iso 1 (557.43 - 570.36 m)

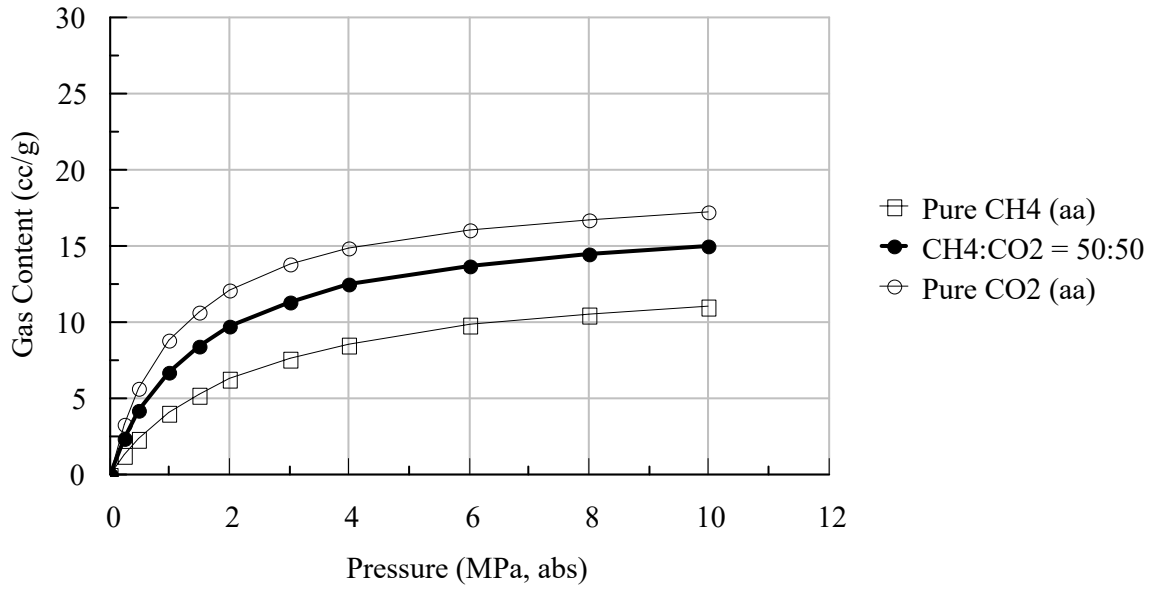
Pure CH4 Langmuir Adsorption Isotherm Coefficients				
	P_L	V_L	P_L	V_L
	(MPa, abs)	(cc/g)	(psia)	(scf/t)
As analysed	2.31	13.67	335	431
daf	2.31	24.28	335	766

Pure CO2 Langmuir Adsorption Isotherm Coefficients				
	P_L	V_L	P_L	V_L
	(MPa, abs)	(cc/g)	(psia)	(scf/t)
As analysed	1.18	19.30	172	609
daf	1.18	34.15	172	1078

Modelled Extended Langmuir Adsorption Isotherm for CH4:CO2 = 50:50					
at 20°C; 101.1kPa (1 atm)			at 60°F, 14.7 psia		
Pressure	Gas Content	Gas Content	Pressure	Gas Content	Gas Content
(MPa)	(cc/g)	(cc/g)	(psia)	(scf/t)	(scf/t)
(absolute)	as analysed	(daf)		(as analysed)	(daf)
0.250	2.40	4.24	36	76	134
0.500	4.21	7.46	73	133	235
1.000	6.78	12.01	145	214	379
1.500	8.51	15.08	218	269	476
2.000	9.76	17.29	290	308	546
3.000	11.43	20.25	435	361	639
4.000	12.50	22.15	580	395	699
6.000	13.79	24.44	870	435	771
8.000	14.54	25.77	1160	459	813

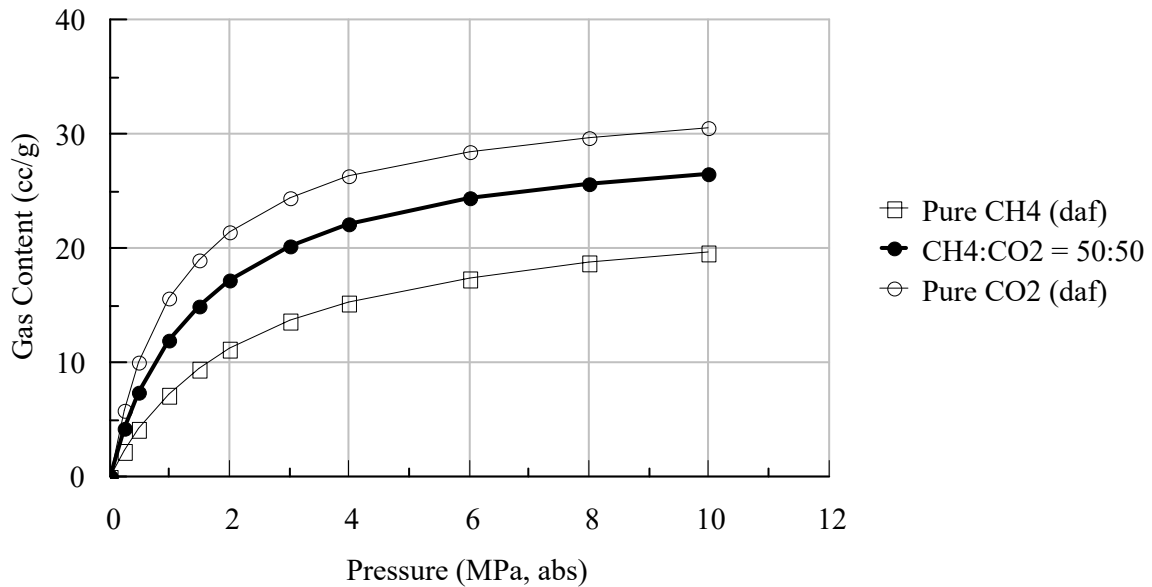
Extended Langmuir Adsorption Isotherm Coefficients				
	P_L	V_L	P_L	V_L
	(MPa, abs)	(cc/g)	(psia)	(scf/t)
As analysed	1.56	17.39	227	549
daf	1.56	30.81	227	972

Extended Langmuir Adsorption Isotherm - CH₄:CO₂ = 50:50 (20 °C, 10 MPa)
 Last Chance #3 - Iso 1 (557.43 - 570.36 m)
 Analysis Temperature 54.5 °C



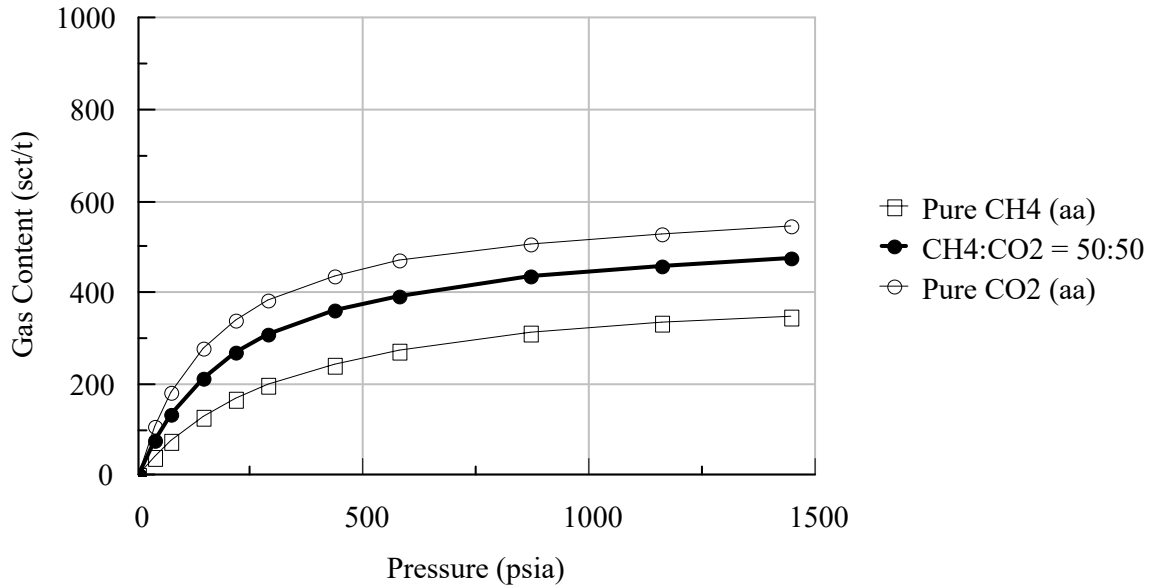
$$V = 17.39 P / (P + 1.56) \quad (\text{As analysed})$$

Extended Langmuir Adsorption Isotherm - CH₄:CO₂ = 50:50 (20 °C, 10 MPa)
 Last Chance #3 - Iso 1 (557.43 - 570.36 m)
 Analysis Temperature 54.5 °C



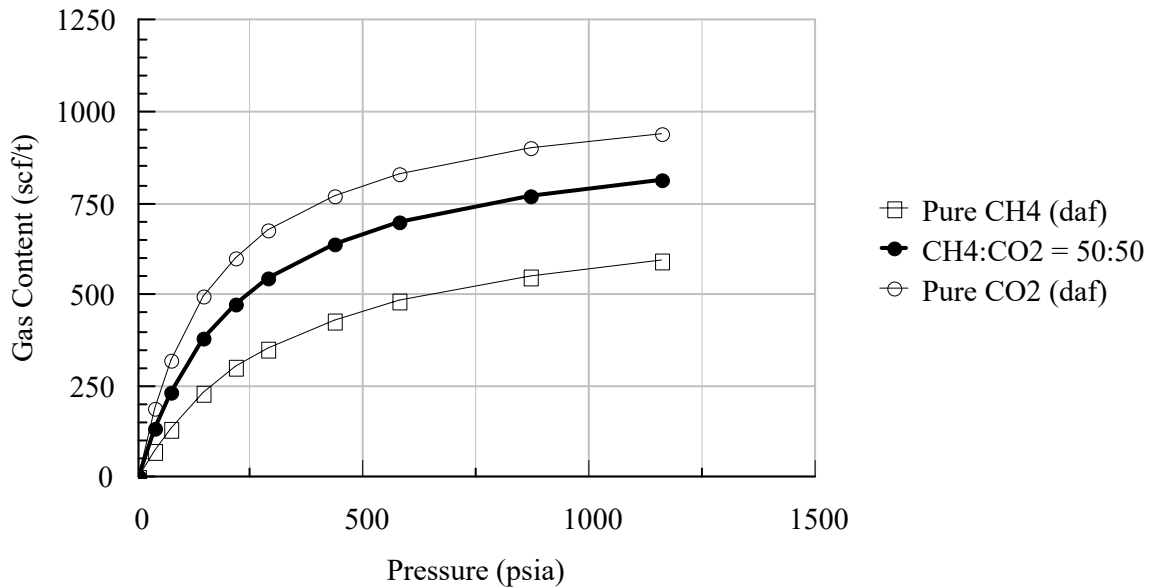
$$V = 30.81 P / (P + 1.56) \quad (\text{daf})$$

Extended Langmuir Adsorption Isotherm - CH₄:CO₂ = 50:50 (60 °F, 14.7 psia)
 Last Chance #3 - Iso 1 (557.43 - 570.36 m)
 Analysis Temperature 130.1 °F



$$V = 549 P / (P + 227) \quad (\text{As analysed})$$

Extended Langmuir Adsorption Isotherm - CH₄:CO₂ = 50:50 (60 °F, 14.7 psia)
 Last Chance #3 - Iso 1 (557.43 - 570.36 m)
 Analysis Temperature 130.1 °F



$$V = 972 P / (P + 227) \quad (\text{daf})$$

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Director, ERC
1st April, 2050